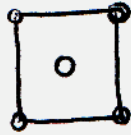


1.9. (a)



$$\# \text{ of atom} = 1 + 4 \times \frac{1}{4} = 2$$

$$\text{lattice constant } a =$$

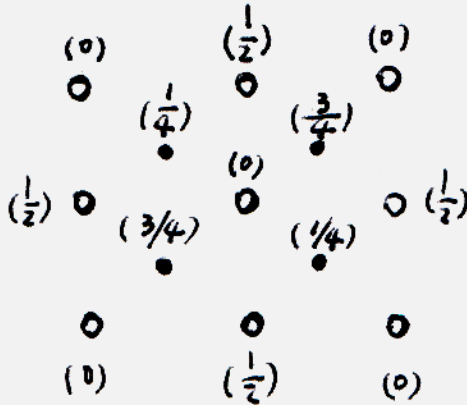
$$\text{surface density} = \frac{2}{a^2} = 6.78 \times 10^{14} \text{ atoms/cm}^2$$

(b) Fcc lattice

The nearest In neighbour distance is

$$\frac{\sqrt{2}}{2} a = \frac{\sqrt{2}}{2} \times 5.87 \text{ \AA} = 4.15 \text{ \AA}$$

1.11.

Top view along
the direction $\langle 100 \rangle$

1.14. packing factor (f)

$$\text{SC} \quad f = 0.52$$

$$\text{BCC} \quad f = 0.68$$

$$\text{Diamond} \quad f = 0.34$$