High-temperature superconductivity of *Pm3n* Lu₄H₂₃ immersed in an as-synthesized lutetium polyhydride

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The precompression effect of foreign elements affords a promising route to metallizing hydrogen atoms under moderate pressure and achieving superconductivity with high transition temperature (T_C) in polyhydrides. Recently, lutetium (Lu) polyhydrides are receiving considerable attention due to their fully filled f orbitals, which are favorable for realizing high T_C . The highest T_C observed in the as-synthesized lutetium polyhydrides up to now has reached 65–71 K under pressures of 181–218 GPa, which was attributed to the $Pm\bar{3}n$ Lu₄H₂₃ phase but without computational analysis. Here we perform first-principles calculations on the bonding feature, the stability, the electronic property, and the superconductivity of Lu₄H₂₃. Lu₄H₂₃ presents robust metallicity due to the orbital hybridizations associated with the H-H covalent and Lu-H ionic bonds, as well as the charge transfer from Lu to H atoms. Our calculations reveal that Lu₄H₂₃ can be stabilized at pressures above ~200 GPa, whose T_C is estimated to be 69–225 (103–210) K at 200 (218) GPa and maintains at the average values around 150–200 K under the pressures lower than 200 GPa. We propose that the $Fm\bar{3}m$ LuH plays a possible role in stabilizing Lu₄H₂₃ at pressures lower than 200 GPa, and, simultaneously, decreases the T_C of superconductivity in the as-synthesized lutetium polyhydrides. Our work provides a computational analysis on Lu₄H₂₃, and the predicted high T_C is expected to be experimentally realized by increasing pressure and optimizing synthesis process to eliminate the LuH phase.

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I. INTRODUCTION

Since first discovered by Onnes in 1911, superconductivity (SC) has received continuous and extensive research attention in the fields of condensed matter physics and material science. Increasing the transition temperature $(T_{\rm C})$ of SC is one ambitious motivation in pursuing novel SC candidates. It was theoretically proposed that metallic hydrogen (H) with high Debye temperature and strong electron-phonon coupling (EPC) is promising in realizing high- $T_{\rm C}$ SC within the framework of the Bardeen-Cooper-Schrieffer theory [1-4]. However, the formation of metallic hydrogen demands extremely high pressure, ~ 500 GPa [5], which is still a great challenge and impractical for application. The chemical precompression effect of foreign elements provides a possible route to decrease pressure [6], as evidenced by discovering the SC of hydrogen sulfide with $T_{\rm C} = 203$ K under 155 GPa [7–10].

Rare-earth polyhydrides with clathratelike structure [11], such as Sc-H [12,13], Y-H [14,15], La-H [16], Ac-H [17], Ce-H [18], Th-H [19], and La-Y-H [20] systems, represent a large material family with high- $T_{\rm C}$ SC. The $T_{\rm C}$ of $Fm\bar{3}m$ LaH₁₀ was initially predicted to be 274–286 K under 210 GPa [21] and then experimentally confirmed with $T_{\rm C} = 250 - 260$ K under 170–180 GPa [22,23]. Subsequent experiments show the $Fm\bar{3}m$ LaH₁₀ can be decompression transformed to C_2/m phase at critical pressures of around 138–160 GPa [24,25], which results in a steep variation of $T_{\rm C}$ below and a slow variation above the critical pressure. The high $T_{\rm C}$ of LaH₁₀ is attributed to the non-neglectable density of states (DOS) of the H atom at the Fermi level $E_{\rm F}$ [26] and the unfilled f shells of the La atom located at the beginning of the lanthanide series [27]. This fact provokes the prediction of high- $T_{\rm C}$ SC in the polyhydrides of later lanthanides with fully filled f shells [28], such as the $Im\bar{3}m$ LuH₆ with $T_{\rm C} = 273$ K at 100 GPa [29] and the *Immm* LuH₆ with $T_{\rm C} = 86.2$ K at 300 GPa [30]. Experimentally, in addition to the identified SC with $T_{\rm C} \sim$ 12.4 K of $Fm\bar{3}m$ LuH₃ at 122 GPa [12], the SC of lutetium polyhydride with $T_{\rm C} = 65 - 71$ K was observed under the pressure of 181-218 GPa [31]. The in situ high-pressure xray diffraction (XRD) experiments imply the as-synthesized lutetium polyhydride containing $Pm\bar{3}n$ Lu₄H₂₃ and $Fm\bar{3}m$ LuH, where the former was proposed to be responsible for the observed SC but without further computational analysis [31].

Computational analysis based on density functional theory provides a powerful tool not only in predicting high- $T_{\rm C}$ polyhydrides, but also in interpreting experimental results. Recently, a paper that falsely reported near-ambient SC with $T_{\rm C} = 273$ K in a nitrogen-doped lutetium hydride (Lu-N-H) was retracted [32], which, on the other hand, led to notable research attention on the fluorite-type LuH₂ with $Fm\bar{3}m$ crystal symmetry. The $Fm\bar{3}m$ LuH₂ was computationally proposed to be the dominant component of the Lu-N-H system based on constructing a Lu-N-H phase diagram as well as simulating optical absorption spectra and XRD patterns

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[33–37], which is in good agreement with the conclusion of related experiments and the observed color changes under pressure [38–42]. Furthermore, the absence of SC under near-ambient pressure in the $Fm\bar{3}m$ LuH₂ was consistently revealed by both computational [35–37] and experimental reports [40–43].

In this paper, we provide a computational insight into the bonding feature, the stability, the electronic property, and the SC of $Pm\bar{3}n$ Lu₄H₂₃. Our calculations indicate that the $Pm\bar{3}n$ Lu₄H₂₃ can be stabilized under pressures higher than 200 GPa, and the stable $Fm\bar{3}m$ LuH phase plays the role of stabilizing the $Pm\bar{3}n$ Lu₄H₂₃ in the as-synthesized lutetium polyhydride under pressures lower than 200 GPa [31]. The metallicity of the $Pm\bar{3}n$ Lu₄H₂₃ stems from the orbital hybridizations and the charge transfer between the Lu and H atoms, which enables the precondition of emerging SC with high $T_{\rm C}$. The estimated $T_{\rm C}$ is 103–210 K at 218 GPa and changes to 69-225 K at 200 GPa, where the low limit of $T_{\rm C}$ approaches the experimentally reported $T_{\rm C} = 65 - 71$ K at 181–218 GPa in the as-synthesized lutetium polyhydride [31]. We attribute the non-negligible deviation of $T_{\rm C}$ to the existence of the $Fm\bar{3}m$ LuH phase with low $T_{\rm C}$ in the assynthesized samples. The average $T_{\rm C}$ of Lu₄H₂₃ is around 150-200 K when the pressures exceed 250 GPa, which is expected to be experimentally realized by optimizing the synthesis process to eliminate the LuH phase.

II. RESULTS AND DISCUSSIONS

A. The bonding feature, stability, and electronic property of Lu₄H₂₃

The $Pm\bar{3}n$ Lu₄H₂₃ adopts the structure of Weaire-Phelan foam, which is stacked by the clathrates with 12 or 14 facets [Fig. 1(a)]. The clathrate with 12 facets is constructed by 20 H atoms and the facets are all pentagons, while the clathrate with 14 facets consists of 24 H atoms and the 14 facets include 2 hexagons and 10 pentagons. The 20 H clathrates are separated by the 24 H clathrates, and the 24 H clathrates connect with each other by sharing edges or hexagonal facets. Each clathrate encases one Lu atom [Fig. 1(a)]. The Lu atoms within the 20 H clathrates occupy the corner or center of the $Pm\bar{3}n$ Lu₄H₂₃ primitive cell, and the Lu atoms within the 24 H clathrates locate at the faces of the Lu₄H₂₃ primitive cell. The bonding feature, the stability, and the electronic property of Lu₄H₂₃ were investigated by using the QUANTUM ESPRESSO (QE) package [44] and the Vienna Ab initio simulation package [45], where the computational details are presented in Sec. I of the Supplemental Material (SM) [46], which includes Refs. [31,44,45,47-60].

The crystal structure of the $Pm\bar{3}n$ Lu₄H₂₃ was fully optimized under different pressures and our calculated results indicate the van der Waals interaction is necessary to reproduce the experimentally reported lattice constant *a* of the $Pm\bar{3}n$ Lu₄H₂₃ at 185 GPa [upper panel of Fig. 1(b)], which gradually decreases with the increase of pressure. The bonding feature is investigated by analyzing the H-H bond lengths (*d*_H), the distribution of electron localization function (ELF), and the crystal orbital Hamiltonian population (COHP). Most of the H-H bonds in Lu₄H₂₃ have lengths shorter than



FIG. 1. (a) The crystal structure of $Pm\bar{3}n$ Lu₄H₂₃. (b) Upper panel: the calculated lattice constants *a* of Lu₄H₂₃ under different pressures, where the red star denotes the experimentally reported lattice constant at 185 GPa. Lower panel: the variation of H-H bond lengths *d*_H under different pressures, and the number of bonds with a given length is drawn proportionally to the magnitude of the circle. (c) The ELF distributions of Lu₄H₂₃ at 218 GPa, where the value of ELF on the cutting planes is represented by color and the isosurface of ELF is set to 0.5. The H-H bonds with the length being longer than 1.3 Å and shorter than 1.1 Å are respectively marked by upward and downward arrows, while the H-H bonds without marked arrows have the length between 1.1 and 1.3 Å. (d) The calculated COHP for the H-H and Lu-H bonds of the $Pm\bar{3}n$ Lu₄H₂₃ at 218 GPa.

1.3 Å but longer than 1.1 Å [lower panel of Fig. 1(b)], which present weak covalent bonding interaction as indicated by the associated ELF value of ~ 0.6 [Fig. 1(c)]. The H-H bonds with lengths shorter than 1.0 Å connect the 20 H clathrates along the diagonal direction of the Lu₄H₂₃ primitive cell, which is a strong covalent bond with an ELF value exceeding 0.8 [Fig. 1(c)]. The length of H-H bonds longer than 1.3 Å corresponds to two parallel edges of the shared hexagonal facets between 24 H cages, which have ELF values smaller than 0.5 [Fig. 1(c)]. The ELF value between the Lu and H atoms is close to zero [Fig. 1(c)], suggesting their interactions are ionic bonding in nature. The COHP analysis [Fig. 1(d)] reveals that the states between H atoms are antibonding at the Fermi level as indicated by a positive COHP value, while the bonding states at the Fermi level between the Lu and H atoms are verified by a negative COHP value.

The dynamical stability of the $Pm\bar{3}n \operatorname{Lu}_4H_{23}$ under the pressures up to 400 GPa was checked by calculating phonon spectra, which demonstrates the $Pm\bar{3}n \operatorname{Lu}_4H_{23}$ is stable at pressures above 200 GPa, and a lower pressure will lead to imaginary frequency in acoustic modes [Fig. 2(a)]. The unstable phonon modes are closely related to the vibrations of Lu atoms along the direction parallel to the axis perpendicular to the shared hexagonal facets between 24 H cages. Specifically, the phonon instability around the Γ point stems from the shear modes [Fig. 2(b)] and that around the *R* point corresponds to the breathing modes [Fig. 2(c)], where the Lu atoms locating at the center of the 24 H cages play dominant roles on the phonon instability. The phonon instability may be the reason for the coexistence of $Pm\bar{3}n \operatorname{Lu}_4H_{23}$ and $Fm\bar{3}m \operatorname{Lu}H$ in the as-synthesized lutetium polyhydride [31]. Phonon spectrum



FIG. 2. (a) Phonon spectrum of the $Pm\bar{3}n \operatorname{Lu}_4\operatorname{H}_{23}$ at 181 (red lines) and 198 (blue lines) GPa. (b), (c) The vibrations of Lu atoms for the phonon modes with imaginary frequency marked by a (b) red and (c) blue star in the phonon spectrum plotted in (a). (d) Phonon spectrum of the $Fm\bar{3}m$ LuH at 0 (red lines) and 1 (blue lines) GPa. (e) The pressure-dependent distance between Lu atoms d_{Lu} marked in (b) and that marked in the inset of (e), which are respectively shown by black squares and red dots.

calculations indicate the $Fm\bar{3}m$ LuH can be stabilized by a pressure higher than 1 GPa [Fig. 2(d)], and the computational details based on the QE package [44] are also presented in Sec. I of the Supplemental Material (SM) [46]. Our analysis [Fig. 2(e)] shows that the distance between the Lu atoms encased by the 24 H cages [d_{Lu} , marked in Fig. 2(b)] is nearly equal to that [marked in the inset of Fig. 2(e)] of the $Fm\bar{3}m$ LuH. This fact may enable synthesizing the $Pm\bar{3}n$ Lu₄H₂₃ from the $Fm\bar{3}m$ LuH, or vice versa, by sharing the Lu atoms with matched distance, and simultaneously, suppressing the phonon instability of the $Pm\bar{3}n$ Lu₄H₂₃ when the pressure is lower than 200 GPa.

Next, the electronic band structure, DOS, and the charge density were calculated to reveal the metallicity of the $Pm\bar{3}n$ Lu_4H_{23} (Fig. 3). The metallic bands stem from the hybridization between H and Lu atoms [Fig. 3(a)], and the hybridized orbitals include the s and p orbitals of H atoms as well as the s, p, and d orbitals of Lu atoms [Fig. 3(b)]. The hybridization leads to seven Fermi surfaces with each Fermi surface possessing the contributions of both Lu and H atoms (Fig. S3 in the SM [46]). Figure 3(c) shows the distribution of partial charge density for the metallic states near the Fermi level. One can clearly see the Lu atoms within the 20 H clathrates hybridize to surrounded H atoms directly, while the hybridization between the rest of the Lu atoms and their surrounding 24 H atoms is indirect and the center of the shared hexagonal facets works as a bridging point. The latter with indirect hybridization may be one reason for the Lu atoms encased by 24 H clathrates emerging phonon instability first and foremost during decompression. The filling of s and porbitals of H atoms with antibonding features is attributed to the charge transfer from Lu to H atoms. Bader charge analysis reveals that each Lu atom donates about one electron and each H atom accepts ~ 0.2 electron, which can be visualized from the distribution of charge density difference between



FIG. 3. (a) The atom-resolved band structure, (b) projected DOS, and (c) the distribution of partial charge density of the $Pm\bar{3}n$ Lu₄H₂₃ at 218 GPa. The size of the red dot and blue circles in the band structure are respectively drawn proportional to the contribution of H and Lu atoms. The partial charge density is plotted for the metallic states with the energy ranging from $E_{\rm F} - 10$ to $E_{\rm F} + 10$ meV. (d) The distribution of the charge density difference between the self-consistent calculated charge density and the superposition of atomic charge density.

the self-consistent calculated results and the superposition of atomic charge density [Fig. 3(d)]. The blue isosurface around the location of the Lu atom indicates the loss of electrons, while the yellow isosurface surrounding the H atoms reveals the gain of electrons.

B. The SC of Lu₄H₂₃

The metallicity of the $Pm\bar{3}n$ Lu₄H₂₃ provides the precondition of emerging phonon-mediated SC, and the nonneglectable DOS of H atoms at the Fermi level is favorable for enhancing $T_{\rm C}$. The SC of the $Pm\bar{3}n$ Lu₄H₂₃ was estimated by performing EPC calculation based on the QE package [44] (see details from Sec. II of the SM [46]). We present the phonon spectra of the $Pm\bar{3}n$ Lu₄H₂₃ under 218 GPa in Fig. 4(a), which shows an obvious frequency gap around 10–20 THz. The phonon DOS $F(\omega)$ indicates that the gap separates the vibration of Lu from that of H atoms [Fig. 4(b)], with the former (latter) occupying a low (high) frequency range. This is attributed to the significant difference between the atomic mass of Lu and H atoms, whose motions tend to decouple with each other. The calculated isotropic Eliashberg function $\alpha^2 F(\omega)$ presents similar peak structures with $F(\omega)$ [Fig. 4(b)]. This indicates that all phonon modes, i.e., the vibrations of both Lu and H atoms, participate in the process of electron-phonon scattering. Notably, one can see the $\alpha^2 F(\omega)$ is significantly larger than the $F(\omega)$ when the frequency is higher than ~20 THz, while the $\alpha^2 F(\omega)$ and $F(\omega)$ are comparable with each other at lower frequency. This indicates that the vibrations of H atoms enable strong electron-phonon



FIG. 4. (a) Phonon spectrum of the $Pm\bar{3}n \operatorname{Lu}_4H_{23}$ at 218 GPa, where the size of the red circles is drawn proportional to the magnitude of the EPC λ_{kv} . (b) The plot of the phonon DOS $F(\omega)$ and the Eliashberg spectral function $\alpha^2 F(\omega)$, where the $F(\omega)$ is resolved into the vibrations of Lu (blue region) and H (light-blue region) atoms. (c) The cumulative frequency-dependent $\lambda(\omega)$, $\lambda^{Lu}(\omega)$, and $\lambda^{H}(\omega)$ of Lu₄H₂₃ at 218 GPa. (d) The variation of N_F , λ , λ^{Lu} , and λ^{H} of Lu₄H₂₃ under different pressures, where the corresponding data are summarized in Table S2 of the SM [46].

interaction, which scatter electrons more significant than those of Lu atoms.

The cumulative frequency-dependent EPC function $\lambda(\omega)$ was then calculated based on the isotropic Eliashberg function $\alpha^2 F(\omega)$: $\lambda(\omega) = 2 \int_0^{\omega} d\omega' \frac{\alpha^2 F(\omega')}{\alpha'}$. The $\lambda(\omega)$ is quickly increased by the vibrations of both Lu and H atoms [black line in Fig. 4(c)], leading to the total EPC $\lambda = 2.66$ for the Lu₄H₂₃ at 218 GPa. The contributions of Lu and H atoms are $\lambda^{Lu} =$ 1.09 and $\lambda^{H} = 1.58$, which are respectively evaluated by calculating the $\lambda^{Lu}(\omega)$ and $\lambda^{H}(\omega)$ from the isotropic Eliashberg function of Lu and H vibration modes solely [Fig. 4(c)]. In attrition to the comparable $\alpha^2 F(\omega)$ and $F(\omega)$ [Fig. 4(b)], the significant contribution of Lu atoms on the λ can be attributed to its low vibration frequency, i.e., phonon softening, because of the ω^{-1} scaling of $\lambda \sim \alpha^2 F(\omega)/\omega$. This fact can be captured from the momentum-resolved EPC $\lambda_{\boldsymbol{q}\nu}$ for a given phonon branch ν [Fig. 4(a)], where the $\lambda_{q\nu}$ on the phonon modes of Lu is conspicuous. The strong EPC from the vibrations of H atoms can be understood in that the $\alpha^2 F(\omega)$ are significantly larger than its $F(\omega)$ [Fig. 4(b)], because most of the H-H bonds have intermediate lengths ranging from 1.1 to 1.3 Å [Fig. 1(b)] [61,62]. Similar analysis on the phonon spectra, the Eliashberg function $\alpha^2 F(\omega)$, and the EPC strength can be applied to the Lu₄H₂₃ at 200 GPa (Fig. S4 in the SM [46]), 250 GPa (Fig. S5 in the SM [46]), 300 GPa (Fig. S6 in the SM [46]), 350 GPa (Fig. S7 in the SM [46]), and 400 GPa (Fig. S8 in the SM [46]).

The calculated EPC λ , λ^{Lu} , and λ^{H} of Lu₄H₂₃ under different pressures are summarized in Fig. 4(d). Clearly, the total EPC λ is suppressed with the increase of pressure, which decreases rapidly, initially starting from 200 GPa, and then slows down when the pressure exceeds ~300 GPa. This trend can



FIG. 5. (a) The temperature-dependent isotropic (red dots) and anisotropic (blue lines) superconducting gap for the Lu₄H₂₃ under 218 GPa. (b) The summary of $T_{\rm C}$ calculated by using different methods for the Lu₄H₂₃ under different pressures, where the yellow background represents a reasonable range for the $T_{\rm C}$ of Lu₄H₂₃. The black dots denote the experimentally reported $T_{\rm C}$. (c) The cumulative frequency-dependent ω_{\log} , $\bar{\omega}_2^{\rm Lu}$, $\bar{\omega}_2^{\rm H}$, and $\bar{\omega}_2$ of the $Pm\bar{3}n$ Lu₄H₂₃ at 218 GPa. (d) The variation of ω_{\log} , $\bar{\omega}_2$, $\bar{\omega}_2^{\rm Lu}$, and $\bar{\omega}_2^{\rm H}$ of Lu₄H₂₃ under different pressures. The data used to plot figures (b) and (d) are summarized in Table S2 of the SM [46].

be understood from the simplified relation of EPC $\lambda = N_F V_{ep}$ [63], where the $N_{\rm F}$ is the DOS at Fermi level and the $V_{\rm ep}$ is the phonon-mediated pairing potential. The $N_{\rm F}$ decreases monotonically with the increase of pressure [Fig. 4(d)], because the reduced distances between H atoms [Fig. 1(b)], between Lu atoms [Fig. 2(e)], and between Lu and H atoms will enhance electron hopping and hence enlarge the band width. The pressure-dependent $N_{\rm F}$ is nearly parallel to the variation of λ for the pressure above 300 GPa, which indicates the EPC λ is mainly determined by the $N_{\rm F}$ while the $V_{\rm ep}$ remains intact in this pressure range. Differently, the increase of EPC λ is more and more rapid than that of $N_{\rm F}$ when pressure decreases from the 300 GPa. This indicates the dominant component that determines the strength of EPC changes from $N_{\rm F}$ to $V_{\rm ep}$. The variations of λ^{Lu} and λ^{H} indicate that the greatly increased EPC at 200 GPa mainly stems from the vibrations of Lu atoms [Fig. 4(d)], due to the associated phonon softening providing an efficient route to enhancing the pairing potential V_{ep} .

Since the total EPC λ of the $Pm\bar{3}n$ Lu₄H₂₃ under the considered pressures is around or even larger than 1.5 [Fig. 4(d)], representing a strong coupling limit, we evaluated the $T_{\rm C}$ of SC in the $Pm\bar{3}n$ Lu₄H₂₃ by self-consistently calculating the leading edge of superconducting gaps at different temperatures, and the computational details based on the EPW (short name for electron-phonon Wannier) code [60] are presented in Sec. III of the SM [46]. For the Lu₄H₂₃ under 218 GPa [Fig. 5(a)], the gaps evaluated from isotropic Migdal-Eliashberg equations are ~40 meV at zero temperature limit, which are gradually suppressed with the increase of temperature and vanish at the critical temperature of $T_{\rm C}^{\rm iso} = 210$ K. The energy distributions of the superconducting gap were

further calculated by solving anisotropic Migdal-Eliashberg equations. The obtained anisotropic gap varies from ~30 to ~50 meV at 50 K and closes at $T_{\rm C}^{\rm aniso} = 200$ K for the Lu₄H₂₃ under 218 GPa [Fig. 5(a)]. Together with the self-consistently calculated gaps under different pressures (Fig. S9 [46]), we summarized the evaluated $T_{\rm C}^{\rm iso}$ and $T_{\rm C}^{\rm aniso}$ in Fig. 5(b), which indicate the Lu₄H₂₃ holds the possibility of superconducting at the temperatures ranging from ~150 to ~200 K.

We further check the $T_{\rm C}$ of SC in the $Pm\bar{3}n$ Lu₄H₂₃ by comprehensively considering the calculated results of the following McMillan-Allen-Dynes (MAD) [64,65] and Gor'kov-Kresin (GK) equations [66].

$$T_{\rm C}^{\rm MAD1} = \frac{\omega_{\rm log}}{1.2} \exp\left[\frac{-1.04(1+\lambda)}{\lambda - \mu^*(1+0.62\lambda)}\right],$$
 (1)

$$T_{\rm C}^{\rm MAD2} = f_1 f_2 T_{\rm C}^{\rm MAD1}, \qquad (2)$$

$$T_{\rm C}^{\rm GK} = \left(\bar{\omega}_2^{\rm H}\right)^{\lambda^{\rm H}/\lambda} \left(\bar{\omega}_2^{\rm Lu}\right)^{\lambda^{\rm Lu}/\lambda} \exp\left(-\frac{1+\lambda}{\lambda-\mu^*}\right).$$
(3)

Here the Coulomb pseudopotential μ^* is set to 0.1. The logarithmically averaged frequency ω_{\log} and the meansquare frequency $\bar{\omega}_2$ are respectively defined by $\omega_{\log} =$ $\exp[\frac{2}{\lambda}\int_0^\infty \frac{\alpha^2 F(\omega)}{\omega} \ln \omega d\omega] \text{ and } \bar{\omega}_2 = \sqrt{\frac{2}{\lambda}}\int_0^\infty \alpha^2 F(\omega)\omega d\omega. \text{ It is known that the } T_C^{\text{MAD1}} \text{ is valid for } \lambda \lesssim 1.5, \text{ and}$ the inclusion of $f_1 = \left[1 + \left(\frac{\lambda}{2.46(1+3.8\mu^*)}\right)^{3/2}\right]^{1/3}$ and $f_2 =$ $\left[1 + \frac{(\bar{\omega}_2/\omega_{\log} - 1)\lambda^2}{\lambda^2 + \left[1.82(1 + 6.3\mu^*)\bar{\omega}_2/\omega_{\log}\right]^2}\right] \text{ makes the } T_{\rm C}^{\rm MAD2} \text{ reliable when}$ $\lambda > 1.5$. $T_{\rm C}^{\rm GK}$ is valid for $\lambda^{\rm H} \approx \lambda^{\rm Lu}$ [66], and the included mean-square frequencies $\bar{\omega}_2^{\rm H}$ and $\bar{\omega}_2^{\rm Lu}$ are calculated respectively from the isotropic Eliashberg function of Lu and H vibration modes solely. Note that the original T_C^{GK} were formulated for the candidates with well-separated optical and acoustic branches like the hydrogen sulfide [66]. The reason for substituting the EPC and the mean-square frequency of optical and acoustic branches by that of Lu and H atoms is that the phonon modes of more than one Lu atom in the $Pm\bar{3}n$ Lu₄H₂₃ primitive cell can be regarded as folded acoustic branches [Fig. 4(a)].

These equations indicate the $T_{\rm C}$ of SC is not only dependent on the strength of EPC, but also on the logarithmically averaged frequency and the mean-square frequency. The cumulative frequency-dependent ω_{\log} , $\bar{\omega}_2$, $\bar{\omega}_2^{Lu}$, and $\bar{\omega}_2^{H}$ for the Lu_4H_{23} under 218 GPa are plotted in Fig. 5(c). One can clearly see the ω_{\log} contributed by the Lu atoms is only 71.9 K, being 11.9% of the total $\omega_{\log} = 606.4$ K. The mean-square frequencies are calculated to $\bar{\omega}_2 = 1473.4$ K, $\bar{\omega}_2^{\text{Lu}} = 161.9$ K, and $\bar{\omega}_2^{\rm H} = 1911.0$ K [Fig. 5(c)], respectively. These indicate the phonon modes of H atoms are more favorable for enhancing $T_{\rm C}$ than that of Lu atoms, consistent with the initial theory of realizing high-T_C SC from metallic hydrogen. Based on similar calculations on Lu₄H₂₃ under 200 GPa [Fig. S4(d) in the SM [46]], 250 GPa [Fig. S5(d) in the SM [46]], 300 GPa [Fig. S6(d) in the SM [46]], 350 GPa [Fig. S7(d) in the SM [46]], and 400 GPa [Fig. S8(d) in the SM [46]], we obtained $\omega_{\log}, \bar{\omega}_2, \bar{\omega}_2^{Lu}$, and $\bar{\omega}_2^{H}$ under different pressures [Fig. 5(d)], which increase with increasing pressure. The T_C^{MAD1}, T_C^{MAD2} , and $T_{\rm C}^{\rm GK}$ are then evaluated and also summarized in Fig. 5(b).



FIG. 6. (a) The comparison between the XRD patterns of assynthesized lutetium polyhydride measured at 185 GPa (upper panel) [31] and the simulated XRD patterns for the fully relaxed $Pm\bar{3}n \operatorname{Lu}_4H_{23}$ (middle panel) and $Fm\bar{3}m \operatorname{LuH}$ (lower panel) under 185 GPa. (b) The variation of λ , ω_{\log} , and $T_{\rm C}^{\rm MAD1}$ of LuH under pressures ranging from 181 to 400 GPa.

Considering the conditions of the employed Eqs. (1)–(3) to be valid, one can expect the $Pm\bar{3}n$ Lu₄H₂₃ to exhibit SC with $T_{\rm C}$ around 150–200 K when pressure is higher than 250 GPa, consistent with the results obtained by solving Migdal-Eliashberg equations. The $T_{\rm C}$ of Lu₄H₂₃ ranges from ~69 (~103) to ~225 (~210) K under the pressure of 200 (218) GPa [Fig. 5(b)], where the low limit approaches the experimentally reported $T_{\rm C} = 65 - 71$ K at 181–218 GPa in the as-synthesized lutetium polyhydride [31]. We highlight the reasonable range of $T_{\rm C}$ for the Lu₄H₂₃ under different pressures by the yellow background in Fig. 5(b).

Lastly, we should point out that our estimated $T_{\rm C}$ for the $Pm\bar{3}n$ Lu₄H₂₃ [Fig. 5(b)] are higher than the experimentally reported 65-71 K at 181-218 GPa in the as-synthesized lutetium polyhydride [31]. Since the SC of hydrides is highly sensitive to their symmetry and stoichiometry [24,25,67,68], the deviation can be attributed to the as-prepared sample containing the $Fm\bar{3}m$ LuH phase. By simulating XRD patterns, we found the peaks of experimentally measured XRD patterns under 185 GPa [31] can be reproduced by our fully relaxed $Pm\bar{3}n$ Lu₄H₂₃ and $Fm\bar{3}m$ LuH under 185 GPa [Fig. 6(a)]. Notably, the XRD peak around $2\theta \sim 8^{\circ}$ is contributed solely by the $Pm\bar{3}n$ Lu₄H₂₃, which is low in the measured XRD pattern. This fact indicates the component of the $Pm\bar{3}n$ Lu₄H₂₃ may be less in the as-synthesized lutetium polyhydride, and the dominant component should be $Fm\bar{3}m$ LuH. The calculated EPC of the $Fm\bar{3}m$ LuH is around 0.5 under pressures ranging from 181 to 400 GPa [Fig. 6(b)] and is unlikely to enable high- $T_{\rm C}$ SC, which is confirmed by the evaluated $T_{\rm C}^{\rm MAD1}$ of being around 2 K at the pressures of 181-218 GPa and being lower than 4.5 K when the pressure increases from 218 to 400 GPa [Fig. 6(b)]. Consequently, one can expect to improve the $T_{\rm C}$ of lutetium polyhydrides by increasing the component of Lu₄H₂₃ and eliminating the LuH phase starting from Ref. [31].

III. CONCLUSION

In this paper, we have performed comprehensive firstprinciples calculations on the bonding feature, the stability, the electronic property, and the SC of $Pm\bar{3}n$ Lu₄H₂₃ and $Fm\bar{3}m$ LuH. Our calculations indicate that both a H-H covalent bond and a Lu-H ionic bond exist in the $Pm\bar{3}n$ Lu₄H₂₃ with clathrate-like structure, while the $Fm\bar{3}m$ LuH only possesses Lu-H ionic bonds. The different bonding features make the $Pm\bar{3}n$ Lu₄H₂₃ stabilize under pressures higher than 200 GPa, but the LuH is stable once the pressure exceeds ~ 1 GPa. We propose that the LuH plays the role of stabilizing the $Pm\bar{3}n$ Lu₄H₂₃ in the as-synthesized lutetium polyhydride under pressures lower than 200 GPa [31], because the distance between the Lu atoms with phonon instability in Lu₄H₂₃ is well matched to that in LuH. The metallicity of the $Pm\bar{3}n$ Lu₄H₂₃ stems from the hybridization between the orbitals of H and Lu atoms, and the charge transfer from Lu to H atoms fills in the antibonding states between H atoms.

The SC of the $Pm\bar{3}n$ Lu₄H₂₃ is computationally revealed by calculating the EPC strength and estimating $T_{\rm C}$. The EPC strength increases with the decrease of pressure, because the decompression softens phonon modes; it especially softens that of Lu vibrations. The $T_{\rm C}$ is estimated to be 69–225

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(103–210) K under the pressure of 200 (218) GPa, where the low limit approaches the experimentally reported $T_{\rm C} =$ 65 – 71 K at 181–218 GPa in the as-synthesized lutetium polyhydride [31]. We attribute the non-negligible deviation of $T_{\rm C}$ to the presence of the $Fm\bar{3}m$ LuH phase with low $T_{\rm C}$ in the as-synthesized samples. The $T_{\rm C}$ of Lu₄H₂₃ is around 150–200 K at pressures ranging from 250 to 400 GPa, which are expected to be realized by increasing the component of Lu₄H₂₃ and eliminating the LuH phase through optimizing the synthesis process reported in Ref. [31].

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